Sensitivity of microflora to various antibiotics in subacture and chronic inflammation to the maxillary sinuses. Zhur. ush. nos. i gorl. bol. 21 no.4:41-45 Jl-4g '(1. (MIRA 15:1)) 1. Iz kafedry bolezney ukha, gorla i nosa (zav. - dotsent D.A.Bytchenko) Chernovitskogo meditainskogo instituta. (BACTERIA, EFFECT OF IRIGS ON) (SINUSITIS) (ANTIBIOTICS)

TARASYUK, V.Z., kand.med.nauk

Inflatable bougie from a polyeth lene film with a biological membrane for the treatment of acute scalds of the esophagus. Zhur.ush.,nos. i gor.bol.22.no.6:69 N-D*62. (MIRA 16:7)

1. Iz otorinolaringologicheskoy kliniki (zav.-prof. A.A.Gladkov) Chernovitskogo meditsinskogo instituta. (BOUGIES) (ESOPHAGUS—WOUNDS AND INJURIES)

Improvement of corrected services is an important tank of trade workers. Sov. torg. 35 no.11:26428 H '61.

(Retail trade)

KEMULA, W.; KUBLIK, Z.; TARASZEWSKA, ...

Application of the hanging mercury drop electrode to the investigation of anodic passivation of mercury. Bul chim PAN 8 no.5:269-274 160. (EEAI 10:9/10)

1. Institute of Physical Chemistry, Polish Academy of Sciences. Presented by W. Kemula.

(Mercury) (Electrodes, Dropping mercury)

KUBLIK, Z.; TARASZEWSKA, J.

Influence of Clo, No, and So, ions on the properties of the passive mercury electrode. Bul chim PAN 10 no.9:515-520 '62.

1. Institute of Physical Chemistry, Polish Academy of Sciences, and Department of Inorganic Chemistry, University, Warsaw.

KEMULA, Wiktor; KUBLIK, Zenon; TARASZEWSKA, Joanna

Electrolytic accumulation and determination of small amounts of CL, Br, and J ions by cathodic stripping. Chem anal 8 no.2:171-178 '63.

1. Department of Inorganic Chemistry, University, Warsaw, and Institute of Physical Chemistry, Polish Academy of Sciences, Warsaw.

TARASZKIEWICZ, Franciszek

Creatine kinase and its clinical importance. Pol. arch. med. wewmet. 35 no.5:717-721 165.

1. Z Kliniki Chorob Dzieci Akademii Medycznej w Bialymstoku (Kurator: doc. dr. med. B. Bogdanikowa).

TARASZKIEWICZ, Franciszek

Creatine kinase, glutamic-oxaloscetic transaminase and glutamic-pyruvic transaminase activity in the blood serum of children with diphtheritic heart damage. Pediat. Pol. 40 no.5:485-492 My *65.

1. Z Kliniki Chorob Dzieci AM w Hialymstoku (Kierownik: vacat; Kurator: doc. dr. med. B. Bogdanikowa).

TARASZKIEWICZ, Stanislaw; DWORAK, Wlodzim erz

Perirenal abacess caused by Mocardia asteroides. Pol. tyg. lek. 20 no.8:285-286 22 F¹65.

1. Z Kliniki Urologicznej Pomorskiej Akademii Medycznej w Szczecinie (kierownik: doc. dr. A. Wojewski) i z Zakladi Mikrobiologii Pomorskiej Akademii Medycznej w Szczecinie (kierownik: prof. dr. W. Murczynela).

TARASZKIEWICZ, Stanislaw

A case of mucoid adenocarcinoma of the urethra. Pol. przegl. chir. 36 no.4a; Suppl.:623-626 Ap '64.

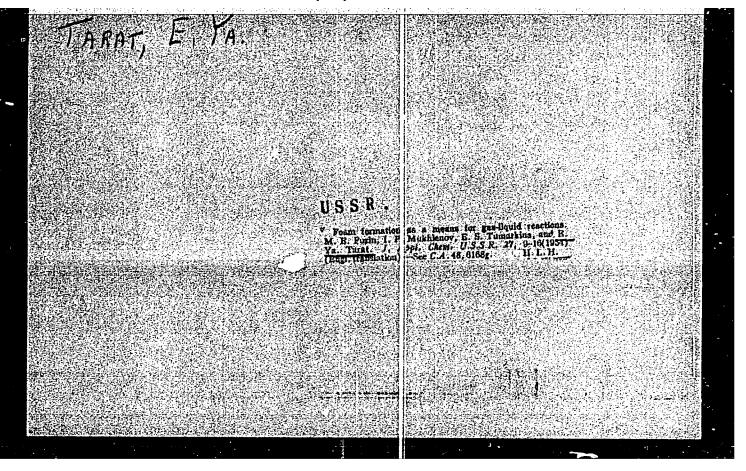
1. Z Kliniki Urologicznej Polskiej Akademii Nauk w Szczecinie (Kierownik: doc. dr A. Wojewski).

TARAT, E. Ia.

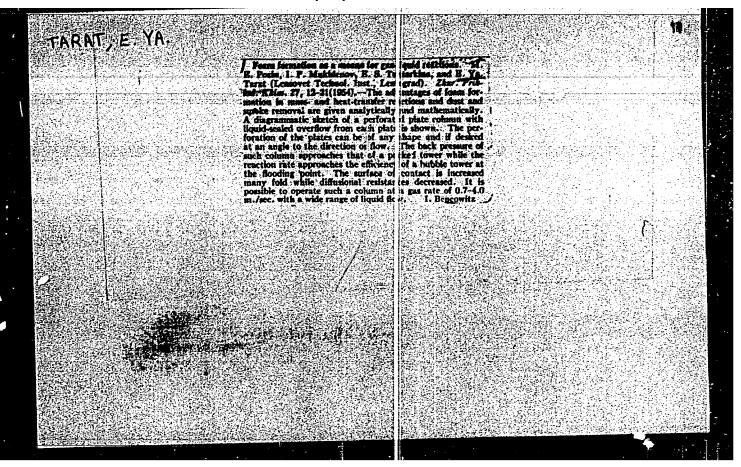
"Investigation of the Hydrodynamic Conditions in the Foam Method of Interacting Gases and Liquids." Cand Tech Sci, Chair of Technology of Monorganic Substances, Leningrad Order of Labor Red Banner Technological Inst imeni Lensovet, Min. Higher Education USSR, Leningrad, 1954. (KL, No 1, 1 Jan 55)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (13) SC: Sum. No. 598, 29 Jul 55

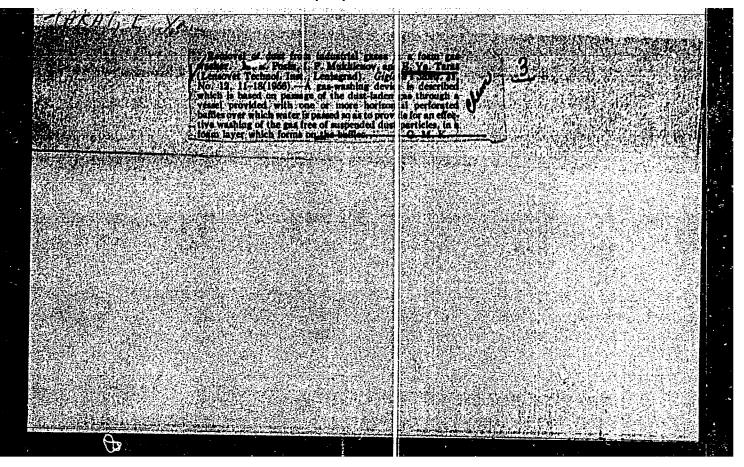
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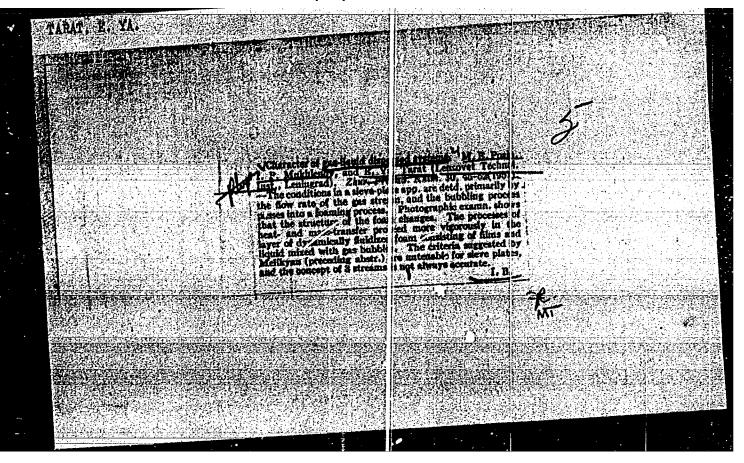
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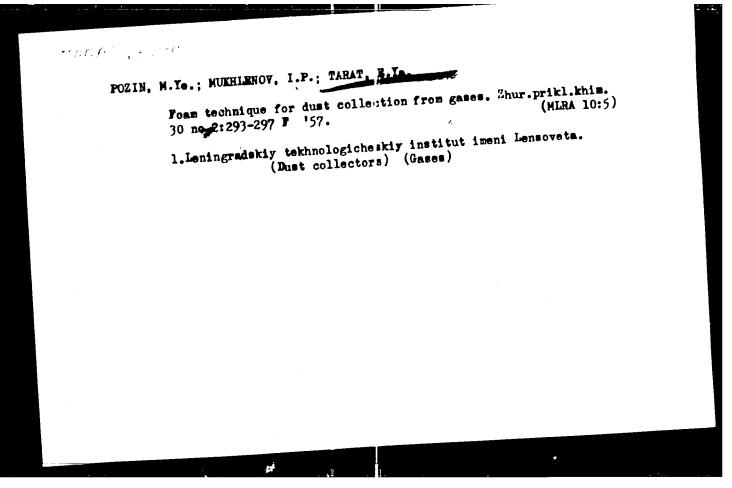


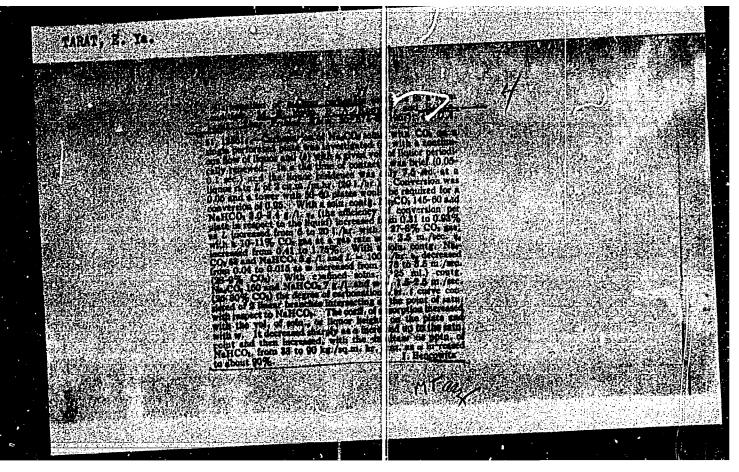
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"APPROVED FOR RELEASE: 07/13/2001 CIA-RDP86-00513R001754930010-1







POZIN, M.Ye.; TARAT, E.Ya.

Absorption kinetics of water vapor by sulfuric acid under turbulent (foam) conditions. Zhur. prik. khim. 31 no.9:1332-1341 S '58. (MIRA 11:10)

1.Leningradskiy tekhnologiche ikiy institut imeni Lensoveta. (Water vapor) (Sulfuric icid) (Absorption)

POZIN, M.Ye.; MUKHLENOV, I.P.; TARAT, E.Ya.; POMKINA, T.A., tekhn.red.

[Froth apparatus for gas Diricication, best and

[Froth apparatus for gas purification, heat exchange, and absorption; operation and calculation for froth apparatus]

Pennye gazoochistiteli, teploobmenniki i absorbery; rabota i raschet pennykh apparatov. Leningrad, Gos.nauchno-tekhn.

izd-vo khim.lit-ry, 1959. 122 p. (MIRA 12:12)

(Gas purification) (Chemical engineering)

14(1)

sov/67-59-3-5/27

AUTHORS:

Pozin, M. Ye., Doctor of Technical Sciences, Professor, Mukhlenov, I. P., Doctor of Technical Sciences, Tarat, E. Ya.,

Candidate of Technical Cciences

TITLE:

On the Height of the Initial Liquid Layer on the Bottom of a Sifting Apparatus (O vysote iskhodnogo sloya zhidkosti na

tarelke sitchatogo apparata)

PERIODICAL: Kislorod, 1959, Nr 3, pp 26 - 31 (USSR)

ABSTRACT:

The height of the initial layer is one of the most important parameters determining the operation of the bottom of a sifting apparatus. The rate of heat- and of mass exchange depends on the height H of the mixture of gas and liquid which forms at the bottom of the sifter (Refs 1,2). H is proportional to the h of the initial height. In this connect-

ion most of the authors do not consider the superelevation of the layer h over the discharge threshold which forms due to

the intensive stream of liquid. In the papers by the authors (Ref 1) it was shown that also without threshold a considerable height H forms due to the stream. Other authors (Aksel'rod,

Card 1/3

On the Height of the Initial Liquid Layer on the Bottom SOV/67-59-3-5/27 of a Sifting Apparatus

Usyukin, and Dil'man, Refs 8,9) assumed only low velocities of the liquid and a constant specific weight of the gasliquid mixture. This charged, however, from 0.1 to almost 1. In this paper a method of determining ho - for apparatus with a discharge device in which h depends on the height of the threshold h_{S} - , on the liquid stream i and on the diameter of the discharge opening, is described. The most simple case is a free discharge without discharge threshold (h depends only on i) a scheme with external discharge is shown on figure 1, a, with threshold and external discharge figure 1,b. 3rd case with consideration of the diameter of the discharge opening figure 1v (H > a + h). In the present investigations two models with a rectangular cross section and with a sifter of the dimensions 500 to 80 and 200 to 60 mm and a variation of the threshold from 0 to 40 mm, and a variation of the discharge opening from 40-120 mm was used. The sifters had circular or slotted openings. The intensity of the liquid stream was varied from 1-75 m37 m hour. The experiments were made

Card 2/3

On the Height of the Initial Liquid Layer on the Bottom SOV/67-59-3-5/27 of a Sifting Apparatus

with air-water of $18-20^{\circ}$ and with increasing temperature also with salt and acid solutions. Moreover, also the formulas for the determination of h (equations 1-13) are developed. The figures show the individual dependences in the variation of different parameters. h may be computed on general practical conditions according to formula $h_0 = \psi h_S + 3 (12)$, mm (8). ψ and ξ may be determined from a companion of the same statement of the same stateme

rison of the data of the two types of apparatus. A more general computation of h is then carried out which may be used for all gas-liquid systems in using different apparatus with a foam formation method (Equations 9-13). From this the equation for h was found:

 $h_0^{0.6} = 1.24 \text{ H/w}^{0.5}$, m (13) where w denotes the velocity of gas. There are 7 figures and 12 references, 11 of which are Soviet.

Card 3/3

sov/80-32-5-14/52

5(2)

AUTHORS:

Pozin, M.Ye., Kopylev, S.A., Tarat, E.Ya.

TITLE:

The Absorption of Carbon Dioxide by Solutions of Caustic Soda Under

Foaming Conditions

PERIODICAL:

Zhurnal priklednoy khimii, 1959, Vol 32, Nr 5, pp 1011-1016 (USSR)

ABSTRACT:

The absorption rate of CO; by a NaOH solution depending on the CO2 concentration in the gasecus phase, the temperature, the linear gas speed, the intensity of the liquid flow, etc. is studied here. The method of calculating the absorption coefficient K and the efficiency factor η is described in Refs 1-7. It has been shown that with the increase of the NaOH concentration from 0.25 n to 2 n the values K and η also increase. At a carbonization degree of 25% of the NaOH solution the decrease of (and η is only slight, but at a higher carbonization degree they decrease rapidly. This is explained by the reduction of the chemical capacity of the solution and by the increase of the resistance to diffusion in the liquid phase due to the appearance of a carbonate layer. The temperature optimum for the absorption lies between 50-70°C. A higher intensity of the gas the absorption lies between 50-70°C. flow increases the absorption due to the higher foam layer. The gas

Card 1/2

SOV/80-32-5-14/52

The Absorption of Carbon Dioxide by Solutions of Coustin Soda Under Foaming

speed, below the value 2.5 m/sec, increases the absorption. At a higher speed the rate of absorption increase is reduced because of the short time of contact. Highly turbulent foaming conditions are very favorable for the process showing results which are 4-5 times higher than in packed column s.

There are: 9 graphs and 14 references, 11 of which are Soviet, 1 English, 1 American and 1 Japanese.

ASSOCIATION: Leningradskiy tekhnologicheskiy institut imeni Lensoveta (Leningrad Technological Institute imeni Lensovet)

SUBMITTED:

October 3, 1957

Card 2/2

Absorption of sulfur dioxide in a foaming state. Izv.vys.ucheb.
zav.;khim. i khim.tekh. 3 no.3: (89-493 *60. (MIRA 14:9)

1. Leningradskiy tekhnologiche:kiy institut imeni Lensoveta,
kafedra tekhnologii neorganiche:kikh veshchestv.
(Sulfur dioxide) (Absorption)

BOGATTKH, S.A., kand.tekhn.nauk; TARAT, F.Ya., kand.tekhn.nauk

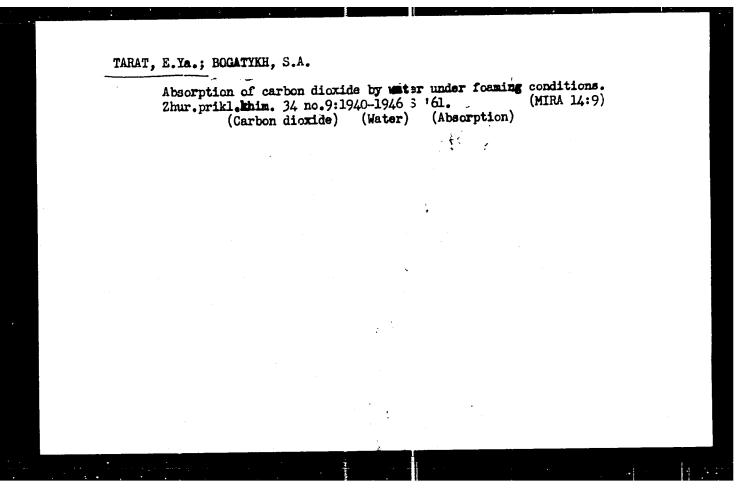
Hydrodynamics of a froth layer ir units with a controlled overflow outlet. Khim. mash. no. 1;14-16 Ja-F'61. (MIRA 14;1)

(Plate towers—Fluid dynamics)

TARAT, E.Ya.; BOGATYKH, S.A.

Absorption of slightly soluble gases in a foam layer. Zhur. prikl.khim. 34 no.8:1886-1889 Ag '61. (MIRA 14:8)

(Absorption)



POZIN, M.Ye.; TARAT, E.Ya.; OREKHOV, I.I.

Ammonium absorption from coke gas in a bubble type apparatus.

Koks i khim. no.9:36-40 '62. (MIRA 16:10)

1. Leningradskiy tekhnologicheskiy institut im. Lensoveta. (Packed towers) (Ammonium) (Coke gas)

POZIN, M.Ye.; TARAT, E.Ya.; OREKHOV, I.I.

Intensification of ammonia distillation from weak ammoniacal liquor. Kosk i khim. no.12:35-40 '63. (MIRA 17:1)

1. Leningradskiy tekhnologicheskiy institut im. Lensoveta.

POZIN, M.Ye.; TARAT, E.Ya.; MRNYAK, L.

Leaking of liquid through the openings of a grid tray in a foam apparatus. Izv.vys.ucheb.zav.;khim. i khim.tekh. 6 no.2:310-319 '63. (MIRA 16:9)

1. Leningradskiy tekhnologicheskiy institut imeni Lensoveta, kafedra tekhnologii neorganicheskikh veshchestv. (Plate towers)

POZIN, M.Ye.; TARAT, F.Ya.; MRNYAK, L.

Height of the initial layer of liquid on a grid tray of a foam apparatus. Izv. vys. ucheb. zav.; khim. i khim. tekh. 6 no.3: 485-489 '63. (MIRA 16:8)

1. Leningradskiy tekhnologicheskiy institut imeni Lensoveta, kafedra tekhnologii neorganiches ikh veshchestv.

(Pate towers)

POZIN, M.Ye.; TARAT. E.Ya.; MRNYAK, L.

Height of the initial layer of liquid on a tray of a foam apparatus with downcomers. Izv. vys. ucheb. zav.; khim. i khim. tekh. 6 no.3:490-497 '63. (MIRA 16:8)

1. Leningradskiy tekhnologicheski; institut imeni Lensoveta, kafedra tekhnologii neorganicheskikh veshchestv.

(Plate towers)

POZIN, M.Ye.; ZUEOV, V.V.; TERESHCHENKO, L.Ya.; TARAT, E.Ya.; PONCMAREV, Yu.L.

Solubility of nitric oxide in aqueous solutions of some salts. Izv. vys.ucheb.zav.;khim.i khim.tekh. 6 no.4:608-616 '63. (MIRA 17:2)

l. Leningradskiy tekhnologicheskiy institut im. Lensoveta. Kafedra tekhnologii neorganicheskikh veshchestv.

POZIN, M.Ye.; TARAT, E.Ya.; ZUBOV, V.V.; TERESHCHENKO, L.Ya.

Rate and mechanism of absorption of nitrogen oxide by aqueous solutions of salts. Izv.vys.ucheb.zav.; khim. i khim. tekh. 6 no.6:974-981 '63. (MIRA 17:4)

1. Leningradskiy tekhnologicheskiy institut imeni Lensoveta, kafedra tekhnologii neorganciheslikh veshchestv.

MUKHLENOV, I.P.; POZIN, M.Ye.; TARAT, E.Ma.; AZHEL', I.Ya.; VOL'FKOVICH, S.I.;
KUSKOV, V.K.

Bibliography. Zhur. prikl. khiz. 36 no.12:2788-2792 D'63.
(MIHA 17:2)

POZIN, M.Ye.; TARAT, E.Ya.; MORARIU, 1.

Absorption rate of carbon dioxide by monoethanolamine under foaming conditions. Izv.vys.ucheb.zav.; khim. i khim.tekh. 7 no.2:240-245 (MIRA 18:4)

l. Leningradskiy tekhnologich skiy institut im. Lensoveta, kafedra tekhnologii neorganich skikh veshchestv.

BOGATYKII, Semen Aleksandrovich; TARAT, E.Ya., kand. tekhn. nauk, nauchn. red.; SMIRNOV, Yu.I., 1ed.

[Complex air-conditioning in a foam equipment system]

Kompleksnaia obrabotka vozdukhu. v pennykh apparatakh. Leningrad, "Sudostroenie," 1964. 315 p. (MIRA 17:4)

POZIN, M.Ye.; TARAT, E.Ya.; MCRARIU, I.

Height of the initial layer of highly foaming liquids on the grids of foam scrubbers. Izv.vys.ueheb.zav.; khim.i khim.tekh. 7 no.6:1003-1009 *64. (MIRA 18:5)

l. Leningradskiy tekhnologic meskiy institut imeni Lensoveta, kafedra tekhnologii meorgani meskikh veshchestv.

POZIN, M.Ye.; TARAT, E.Ya.; OREKHOV, 1.1.

Efficiency of mass-exchange apparatus as a function of hydrodynamic, physicochemical, and structural parameters. Zhur. prikl. Zhim. 37 nc. 6:1292-1301 Je '64. (MIPA 18:3)

1. Leningradskiy tekhnologicheskiy institut imeni Lencoveta.

POZIN, M.Ye., doktor tekhn.nauk; TARA'. E.Ya., kand.tekhn.nauk; OREKHOV, I.I., kand.tekhn.nauk; TERESHCHENKO, L.Ya., kand.tekhn.nauk

Calculating the efficiency of the shelves of frothers for absorption and desorption processes. Khir., i neft. mashinostr. no.9:11-13 S (MIRA 18:10)

POZIN, M.Ye.; TARAT, E.Ya.; TERESHCHENKO, L.Ya.; ZUBOV, V.V.; TREUSHUHENKO, N.H.

Kinetics of nitrogen oxide absorption with aqueous salt solutions. Izv.vys.ucheb.zav.; khim.i khim.tekh. 8 no.4:628-632 '65. (MIRA 18:11)

1. Leningradskiy tekhnologicheskiy institut imeni Lensoveta, kafedra tekhnologii neorganicheskikh veshchestv.

RUSESCU,A., prof.; BALABAN, I., dr.; POFESCU, Val., dr.; TARATESCU, C, dr.

Clinico-radiclogical aspects of abdominal tumors in children.

Pediatria (Bucur.) 13 no.5*409-420 S-0 '64

TARATIN, A. F., Cand. Tech. Sci. (diss) "Investigation of System of Mutally-connecting Fred Channels of Heavy Profile Grinding Machine," Leningrad, 1961, 19 pp (Leningrad Elec. Engr. Inst.) 200 copies (KL Supp 12-61, 275).

TARATIN, A.F., inzh.

Choice of a method for constructing logarithmic frequency characteristics and analysis of the stability of the joint operation of the channels of copying-milling machine. Izv. LETI no.45:250-276 61.

(Milling machines—Numerical control)

VASIL'YEV, D.V., doktor tekhn. anuk, prof., zasluzhennyy deyatel'; TARATIN, A.F., inzh.

Synthesis of a system of mutually coupled channels of a heavy copying and milling machine. Izv. LETI no.47:220-260 '62. (MIRA 16:12)

S/137/60/000/012/020/041 A006/A001

Translation from: Referativnyy zhurnal, Metallurgiya, 1960, No. 12, pp. 136 - 137, # 29142

AUTHORS:

Podkovich, Ye.G., Taratin, V.

TITLE:

Structure and Properties of Pseudoalloys Obtained by Gas-Flame

Metallizing

PERIODICAL:

Tr. Rostovsk.-n/D. in-ta s-kh. mashinostr., 1959, No. 12, pp. 46-

51

TEXT: An investigation was made of steel and Cu pseudoalloys applied by the methods of gas-flame metallizing and electrometallizing. Steel (0.14% C) and Cu wire, 1.3 mm in diameter was used as initial coating material. Microstructures of pseudoalloys obtained by gas flame and electric metallizing are distinguished by particle dimensions of striated orientation, dimensions and arrangement of oxide films, size and amount of pores. The possibility of regulating the heat source and the protective effect of gas combustion products in gas flame metallizing assure greater density, lesser particle size and a lesser amount of pores in

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S/137/60/000/012/020/041 A006/A001

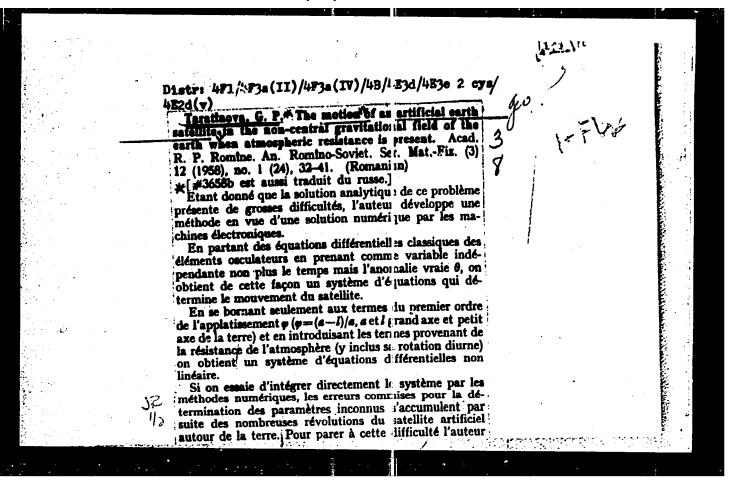
Structure and Properties of Pseudoalloys Obtained by Gas-Flame Metallizing

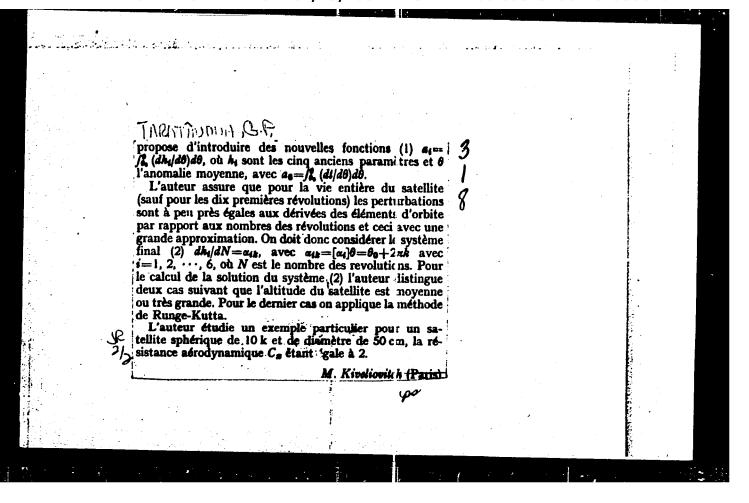
the pseudoalloy, than in electric metallizing. Physico-mechanical properties of pseudoalloys are higher in gas-flame metallizing than in electrometallizing. The coefficient of metal utilization and the uniformity of the particle spraying are also higher in gas flame metallizing. The advantages of gas flame metallizing over electrometallizing consist in the use of simpler devices, higher quality of coatings and lower costs (in particular when replacing C2H2 by natural gas).

S.U.

Translator's note: This is the full translation of the original Russian abstract.

Card 2/2





AFANAS'YEV, A.A.; SLUTSKIY, S.B.; TOLOGIKO, V.I.; Prinimali uchastiye:

KRASHOPOL'SKIY, G.G., inzh.; TAIATHSKIY, M.G., inzh.; TEFLITSKAYA,

K.O., inzh.

Using pig insole leather for so:k lining of Russian leather foot
wear. Kozh.-obuv.prom. 3 no.7:18-21 J1 '61. (MIRA 14:9)

(Shoe manufacture) (Leather)

